

NOVEL N-COORDINATED ORGANOTELLURENIUM AND ORGANOTELLURONIUM CATIONS

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Despite the fact that there are already known some representatives of organotellurenum and organotelluronium cationic species,^[1-8] so far, their reactivity has not been studied in detail. During our look into the interesting reactivity of these ionic compounds we stumbled upon the ability to activate a B-H bond in carborane cluster.^[9] This was the impulse to extend the ranks of known cationic organometallic compounds with Te(II) as a central atom with various chelating and pincer ligands including their subsequent oxidation yielding corresponding cationic Te(IV) species to further study their reactivity. These compounds show impressive levels of Lewis acidity as judged not only by highly deshielded values of δ in ¹²⁵Te NMR, but also by the Gutmann-Beckett method. These results together with outcomes of their extreme tendency to hydrolyse will be presented at the conference.

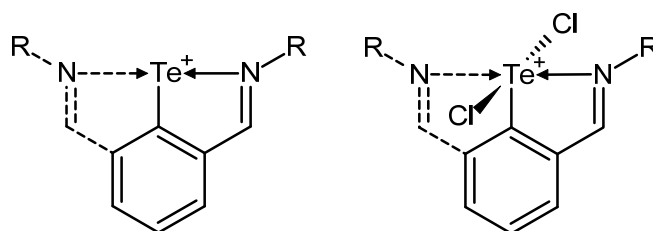


Figure 1. General structures of synthesized cationic compounds.

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