

# COMPUTATIONAL STUDY ON THE REVERSE REACTION OF FORMATE OXIDATION BY MOLYBDENUM-CONTAINING FORMATE DEHYDROGENASE

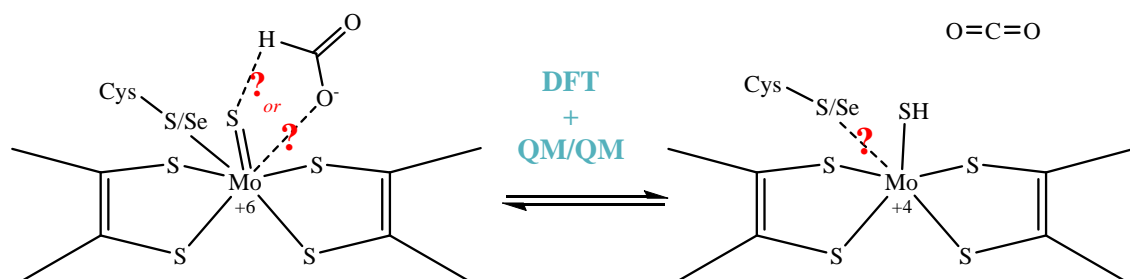
Yanan Han<sup>a</sup>, Olga S. Bokareva<sup>a,b</sup>, Ricardo Mata<sup>c</sup> and Milica Feldt<sup>a</sup>

<sup>a</sup>Leibniz Institute for Catalysis (LIKAT), Albert-Einstein-Str. 29A,  
18059 Rostock, Germany

<sup>b</sup>Institute of Physics, University of Kassel, Heinrich-Plett-Str. 40,  
34132 Kassel, Germany

<sup>c</sup>Institute of Physical Chemistry, Tammannstr. 6, 37077 Göttingen, Germany

As a biological catalyst, Molybdenum-containing formate dehydrogenases (Mo-FDHs) plays an essential role in the formate oxidation. Nowadays, due to the dramatic increase in atmospheric CO<sub>2</sub> levels, more and more experimental and computational studies are turning to CO<sub>2</sub> reduction and effective mechanisms for carbon capture. One of the pathways for the latter is the inverse reaction of formate oxidation [1, 2]. However, the underlying mechanism is still under debate, namely whether the formate directly connects to Mo or the SeCys ligand dissociates from the metal center. In this work, a large QM region of two selected enzymes is chosen to obtain detailed information on the Mo surroundings, a model used previously by Ryde and coworkers [3]. First, we study the full mechanism of the forward reaction of Mo-FDHs. Ligands modifications during the reaction and their contribution to the potential energy are addressed. Further, based on the results for the forward reaction, interactions between substrate and Mo and the ligands change will be investigated.



**Figure 1.** Mechanisms of reverse and forward reactions catalyzed by Mo-FDHs.

[1] Maia, L. B., Moura, I., and Moura, J. J., Environment and Energy, 29-81 (2021).

[2] Siegbahn, P. E., J. Phys. Chem. B, 126, 1728 (2022).

[3] Dong, G., and Ryde, U., J Biol Inorg Chem, 23, 1243 (2018).